

## **PHYSICO-CHEMICAL ANALYSIS OF UNDERGROUND WATER OF JALGAON JAMOD REGION OF BULDANA DISTRICT, MAHARASHTRA, INDIA**

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### **ABSTRACT**

*Ground water samples were collected from different villages of Jalgaon-Jamod (India). These water samples from 25 sampling points of Jalgaon-Jamod were analyzed for their physicochemical characteristics. Laboratory tests were performed for the analysis of samples for pH, Hardness, Chloride, Alkalinity, TDS etc. On comparing the results against drinking water quality standards laid by Indian Council of Medical Research (ICMR) and World Health Organization (WHO), it is found that some of the water samples are non-potable for human being due to high concentration of one or the other parameter. The usefulness of these parameters in predicting ground water quality characteristics were discussed. Thus an attempt has been made to find the quality of ground water in and around Jalgaon-Jamod, suitable for drinking purposes or not.*

**Key Words:** *Water quality, physiochemical parameters, pollution study, drinking water. Hariharataluk.*

### **INTRODUCTION**

Ground water is crucial, most suitable fresh water resource with nearly balanced concentration of the salts for human consumption. Surface water and ground water is the only available source for human being but due to industrializations, urbanization and agriculture practices the surface water resources are greatly polluted. Relatively groundwater resources are much clean and usable water resources.

Today a large number of rural and urban populations depend upon groundwater. It is estimated that approximately one third of world's population depends upon ground water resources for drinking (Nickson et al. 2005). The greater part of state in India where more than 90% of population depends upon ground water for drinking (Ramachandraiah 2004). Over burden of the population pressure, unplanned urbanization, unrestricted exploration policies and dumping of the polluted water at inappropriate place enhance the infiltration of harmful compounds to the ground water. Ground water occurs in weathered portion, along the joints and fractures of the rocks. In fact, industrial waste and the municipal solid waste have emerged as one of the leading cause of pollution of surface and ground water. In many parts of the country available water is rendered non-portable because of the presence of heavy metal in excess. The situation gets worsened during the summer season due to water scarcity and rain water discharge. Contamination of water resources available for household and drinking purposes with heavy metals, metal ions and harmful microorganisms is one of the serious major health problems.

In the Jalgaon-Jamod ground water is the main source of fresh water for major population.

Groundwater is usually extracted through hand dug shallow wells. Selected places of Jalgaon-Jamod been taken for the present study. It lies adjacent to each other in the north of Buldhana district of Maharashtra.

## **MATERIALS AND METHODS**

**Study Area:** Jalgaon-Jamod situated at the base of Satpura Range, about 20 km from Satpuda and its Coordinates are 21.0486°N 76.5344°E and the total population in Jalgaon-Jamod tahasil is 156,623 as per the survey of census during 2011 by Indian Government. Out of total population 128,347 people are living in the urban area and about 128,347 are living in rural areas. There are 33,861 House Holds in this sub district. There are 80,008 males (51%); There are 76,615 females (49%). It has an average elevation of 291 metres (955 feet). Purna is the largest river in the tehsil which is also largest in the District. There is the satpuda mountain ranges in the northern part of the tehsil.

Sampling location of study area is shown in the Table No:1.1

Sl. No.	Sample Number	Name of Village	Sl. No.	Sample Number	Name of Village
1	DW1	Bhon	14	BW14	Ukalgaon
2	DW2	Pesoda	15	BW15	Pimpari
3	DW3	Khiroda	16	BW16	Rudhana
4	DW4	Savali	17	BW17	Bhilkhed
5	DW5	Paturda	18	BW18	Kathargaon
6	DW6	Awar	19	DW19	Marod
7	BW7	Kavathal	20	BW20	Wakana
8	DW8	WarvatKhanderao	21	DW21	Wankhed
9	DW9	Jastagaon	22	DW22	Bodakha
10	DW10	Kalamkhed	23	BW23	Tamgaon
11	BW11	Takaleshwar	24	BW24	Nirod
12	BW12	Chondi	25	DW25	Ringanwadi
13	BW13	Kakoda			

BW: Borewell, DW: Dugwell

## **Sample collection**

The water samples were collected in pre-cleaned one liter capacity polyethylene containers of 2 litre capacity. The containers were pre-washed with chromic acid solution and rinse with distilled water several times

## **Physico-Chemical Analysis**

The collected samples were analyzed for different physico-chemical parameters such as pH, Electrical conductivity (EC), Total Dissolved solids (TDS), Total hardness (TH), Total Alkalinity and Chloride as per the standard methods (APHA 1998), and the results were compared with the Indian Standards (ICMR) for potable water. The parameters present in the water sample can be calculated by using various methods. The pH of all the water samples was determined using a pH meter (Model no LI 127, Elico) Electrical conductivity was measured using a conductivity meter. The chloride, and total hardness were estimated by the standard methods of water

## **RESULTS AND DISCUSSION**

A total of 25 water samples from hand pumps and bore wells of Jalgaon-Jamod were collected in pre-cleaned one liter capacity polyethylene containers of 2 litre capacity, which were being used mainly for the purpose of drinking and cooking and are brought to the laboratory with proper

care. The water quality analysis of different ground water samples have been carried out for pH, Electrical conductivity, Total Alkalinity, TDS, Total hardness and Chloride. The samples were analysed using various analytical methods, APHA (1998), AWWA-WPCF (1995). Total hardness and calcium were measured by EDTA titrimetric method using EBT indicator respectively. Chloride was determined by Argentometric method using potassium chromate indicator. The chemical data were compiled further to know location wise distribution. The data revealed that there were considerable variations in the examined samples from different sources with respect to their chemical characteristics. The results indicate that the quality of water considerably varies from location to location.

**Hydrogen ion activity (pH):** pH of water is an essential and prime parameters of groundwater. From Table 1.2 reveals that pH values of samples ranged varied 8.15 to 9 in case of post monsoon season and Table 1.3 reveals that pH values of samples ranged between 7.8 to 8.57 for pre-monsoon season.

**Electrical conductivity (EC):** From Table 1.2 shown that groundwater EC values are found in ranged 412 to 7120 in post monsoon session. Table 1.3 shown that groundwater EC values are ranged 312 to 6750 are found in pre-monsoon session.

**Total Alkalinity (TA):** From Table 1.2 shown that Total Alkalinity values are found in ranged 203 to 821 in post monsoon session. Table 1.3 shown that Total Alkalinity values are ranged 160 to 720 are found in pre-monsoon session.

**Total Dissolve Solid (TDS):** 512 to 9556 is the range of TDS in pre-monsoon session and in post monsoon session it varies from 400 to 9984

**Total hardness (TH):** Total hardness value range from 200 to 4412 in pre monsoon session and 112 to 4328 is the range of value for post monsoon session

**Chloride (Cl<sup>-</sup>):** From Table 1.2 shown that chloride values are found in ranged 79 to 836 in post monsoon session. Table 1.3 shown that groundwater EC values are ranged 55 to 759 are found in pre-monsoon session.

## **CONCLUSION**

In general ground water quality of Jalgaon-Jamod is not harmful to human beings, since the ground water which were taken from the various places of Jalgaon-Jamod were analyzed and the analysis reports that the water quality parameters like pH, Cl<sup>-</sup>, Total Alkalinity lies within the maximum permissible limit prescribed by WHO and ICMR. Except certain parameters like Total hardness, TDS, EC etc. As study area comes under the saline belt it is recommend to boil the water before use it minimize the temporary hardness of water.

**Table 1.2: Physicochemical analysis of water from Jalgaon Jamod and Sangrampur region in Post Monsoon- 2015**

**Table 1.3: Physicochemical analysis of water from Jalgaon, Jamod and Sangrampur region in Pre Monsoon- 2016**

S. No.	TA	Cl-	TH	TDS	E.C	pH	S. No.	TA	Cl-	TH	TDS	E.C	pH
DW1	445	355	252	2000	420	7.74	BW15	205	298	724	400	390	7.66
DW2	212	213	944	9874	460	8.57	BW16	550	355	4200	9984	2030	7.73
DW3	480	76	1696	5992	1240	7.9	BW17	160	92	208	400	6750	7.7
DW4	425	333	176	400	410	7.45	BW18	510	73	1520	5665	1460	8.45
DW5	165	146	323	800	6412	7.88	DW19	260	759	270	800	390	7.68

S. No	TA	Cl-	TH	TDS	E.C	pH	S. No	TA	Cl-	TH	TDS	E.C	pH
DW1	521	412	300	2112	523	8.1	BW15	327	325	811	512	458	8.2
DW2	314	300	1000	9526	564	9	BW16	572	412	4284	9556	2145	7.9
DW3	523	112	1721	4131	1320	8.2	BW17	203	175	312	521	7120	7.9
DW4	519	412	213	600	521	7.8	BW18	571	125	1614	4521	1524	8.45
DW5	245	250	428	956	6523	8.1	DW19	381	836	321	956	451	7.9
DW6	328	200	610	2152	852	8.2	BW20	521	112	1002	2714	1254	8.5
BW7	231	128	231	945	1452	8.2	DW21	523	679	4412	3952	4956	8.4
DW8	301	390	545	2314	412	7.9	DW22	247	789	714	2541	561	7.9
DW9	412	79	480	2564	456	8.1	BW23	352	378	1012	925	521	7.9
DW10	421	115	514	1745	845	7.8	BW24	821	215	600	2714	1625	8.3
BW11	326	89	319	1325	654	8.2	DW25	512	542	641	2141	565	7.7
BW12	362	230	200	956	521	8.5	<b>Maxi</b>	<b>821</b>	<b>836</b>	<b>4412</b>	<b>9556</b>	<b>7120</b>	<b>9</b>
BW13	429	724	550	1754	412	8.4	<b>Mini</b>	<b>203</b>	<b>79</b>	<b>200</b>	<b>512</b>	<b>412</b>	<b>7.7</b>
BW14	251	156	345	1721	1652	8.2	<b>Average</b>	<b>408.52</b>	<b>327.4</b>	<b>927.16</b>	<b>2554.16</b>	<b>1515.48</b>	<b>8.15</b>
DW6	215	163	570	2000	750	7.97	BW20	450	69	928	9871	1150	8.37
BW7	185	71	202	800	1320	7.95	DW21	470	597	4328	9880	3850	8.02
DW8	195	322	480	2000	380	7.73	DW22	180	603	662	2400	420	7.72

DW9	300	55	336	2400	312	7.89	BW23	245	303	982	800	450	7.34
DW10	204	97	444	1600	713	7.13	BW24	720	163	534	2600	1580	8.14
BW11	250	69	276	1200	580	7.83	DW25	315	445	586	2000	420	7.54
BW12	204	177	112	800	425	8.11	<b>Maxi</b>	720	759	4328	9984	6750	8.57
BW13	325	684	402	1600	312	7.92	<b>Mini</b>	160	55	112	400	312	7.13
BW14	175	71	276	1600	1578	7.82	<b>Average</b>	313.8	263.56	857.24	3114.64	1368.08	7.8

## REFERENCES

1. APHA.(1998).StandardMethodsfortheexaminationofwaterandwastewater,Pg2-4,29-179. American Public Health Association.
2. Nickson, R.T., J.M. McArthur, B. Shrestha, T.O. Kyaw-Nyint and D. Lowry, (2005) Arsenic and other drinking water quality issues, Muzaffargarh District, Pakistan. Appl. Geochem., 20(1): 55-68.
3. Ramachandraiah, C. (2004) Right to drinking water in India, Centre for Economic and Social Studies: 56
4. Abimbola O. Adetoro, M.S. Lawal and AdetolaJenyo-Oni, *Advances in Applied Science Research*, **2011**, 2 (2):218-226
5. BishtShikha, Patra, B. A., Gupta N. C., AroraSaurabh, and R. A. Singh.(2007) Assessment of Drinking Water Quality of Delhi, India .In:12th ISMAS-WS-2007, March 25-30, Cidade de Goa, Dona Paula, Goa.
6. Garg VK, Chaudhary A, Deepshikha, Dahiya S, *Indian J Environ Prot*, 19(4) (1999), 267- 272
7. Gupta, B. K. and R. R. Gupta. (1999). *Poll. Res.* 18: 523-525.
8. Gupta, D.P., Sunita and Saharan, J.P. **2009**. *Science pub.*1(2):1-5.
9. Harish Babu, K., Puttaiah, E.T. and Vijaya Kumara, **2006**. *Journal of Living World*. 13(1):18-25.
10. IndrajitSen,AjayShandilandV.S.Shrivastava,*AdvancesinAppliedScienceResearch*,**2011**, 2 (2): 161-166
11. Obi, C.N. and Okocha, C.O.**2007**. *J. of Engg and App. Sci.*2(5):920-929.
12. Rajan M. R. and I. Paneerselvam. (2005). *Indian J. Environ. andEcoplan.* Vol. 10, No.3: 771-776.
13. Shyamala, R., Shanthi. M. and Lalitha, P. **2008**. *E-Journal of Chemistry*.5(4):924-929.
14. Thakare S. B., A. V. Parvate and M. Rao. (2005). *Indian J. Environ. andEcoplan.* Vol. 10 No.3:657-661.
15. Yadav S.S. and Kumar Rajesh, *Advances in Applied Science Research*, **2011**, 2 (2):197-201